

Horizontal MOCVD Injector Reactor TNL- Injector Simulator



CHALLENGES: MOCVD EPITAXY

□ MOCVD → *Chemical process* to deposit special materials and high-tech devices Growth mechanism Investigation of III–V compounds is *High value products* □ Film growth properties, closely related to product quality Depends gas *inlet flow rate*, operating pressure & temperature and reactor geometry configurations □ High-quality demands of the semiconductor industry control composition distributions in a MOCVD reactor to achieve *film growth thickness* & *uniformity* □ MOCVD : *complex transport phenomena* → momentum, heat, mass & chemical reactions \Box MOCVD reactor's optimal growth condition \Rightarrow requires extra experimentation to refine & search process to locate an optimal solution until film deposition quality satisfies a prespecified criteria.

Two categories: Development of a chemical kinetic & Transport phenomenon □ To save design cost & development time → use of *CAD methodologies* of MOCVD processes.





CHALLENGES: GaN/Si EPITAXY

large lattice mismatch (16% to GaN directly on Si encounters several challenges 20.4%) and thermal expansion coefficient (TEC) mismatch ~ 53% leads to the generation of defects at room temperature

□ AIN buffer layer growth prior to the growth of GaN _ To avoid formation of cracks and several other technological challenges

 \Box Stack of $Al_xGa_{1-x}N$ layers on AIN buffer layer \Box To balance compressive and tensile strain

Growth of the AIN buffer layer on the Si substrate Still not well understood

• Well known phenomenon of formation of *amorphous SiNx*, deteriorates crystalline quality. Pre-nitridation process of Si substrate greatly influences surface morphology. However, nitridation time and substrate nitridation temperature highly influence strain generation between each layer and results in defect formation process

 \Box NH₃ pre-flow with smaller step size time and the optimum substrate temperature are still *debatable* to achieve a single-crystalline GaN on Si substrate





CHALLENGES: GaN/Si EPITAXY

GaN-on-Si devices recently attracted much attention *for radio frequency* (RF) applications due to cost & large size Si substrates with possibility of co-processing in CMOS foundries However, GaN-on-Si RF devices, still facing some challenges in form of *RF loss* Contributor to RF loss is *parasitic channel formation* at AIN/Si interface

- Type of conductivity & formation mechanism of parasitic channel is *controversial* Given the suggest formation of a p-type conductive layer at the AIN/Si interface, while others suggest that an n-type electron channel induced by the strong polarization field at the AIN/Si interface.
- Understanding the behaviors of the parasitic layer is crucial for reducing the RF loss Unsolved problem requires an unambiguous identification of this conductive layer and correlating it with the RF loss







□ Numerical simulations, (CFD software) ONLY predicts → Film growth rate and uniformity, Prediction accuracy heavily relies on the reaction kinetics mechanism, however, transport phenomena oversimplified by assuming constant mass diffusivity for precursors, temperature-independent properties of gas mixture, and constant wall temperature profile or constant heat transfer coefficient for boundary condition of inner reactor wall.

• Oversimplifications made for the heat transfer >> substantially prohibit the prediction ability of the MOCVD processes, making a great discrepancy between the model and the actual behavior of the film growth on substrate.

To extract atomistic scale information of MOCVD reactor process through CFD software is **IMPOSSIBLE**





INPUTS: MOCVD PROCESS



Chamber Condition Showerhead Based Injector Based 	Injector Parameters			Precurssor Condition Number of Port Precursor 1 Flow Rate	0 😴 Select Prec 🔻	
	Chamber Volume (ltrs.) Chamber Pressure Ceiling Height (cm)	1.4 10 2.0 100 1	torr 💌	Load Reaction	Step 1 Load 1	
	Chamber Temperature (C) Sticking Coeff.			Number of Port Precursor 1	4 🗢 Select Prec V Select Precursc	
Many More parameter	s details Require			Load Reaction	Ga(CH3)3 (CH3)3CAsH2 H2 Al(CH3)3 NH3 CH3 O2	





METHODOLOGY: MOCVD PROCESS



Schwoebel barrier: The atom diffuses from the site exactly above the edge atom to the site immediately next to the edge atom as;

Incorporation barrier: incorporates into the edge on the same surface level.









The atom

PLANETARY MOCVD PROCESS

Planetary MOCVD architecture implemented in *TNL-Injector* simulator equivalent to the AIX 200/4 horizontal MOCVD reactor.

Inlet of the reactor is divided into two parts by a separator through which the group III and V precursors can be fed into the upper and lower inlet respectively.









where J_A is the diffusion flux of specie A, is the concentration of species A, x is the direction perpendicular to the substrate surface, *R* is the gas constant, *T* is the absolutc_A e temperature. δ is the chamber boundary layer thickness.

 D_{A_B} is the diffusivity of the bulk stream reactants and dependent on Leonard-Jones parameters (σ, Ω) based on the Chapman-Enskog theory

$$D_{A_B} = 2.7 \times 10^{-3} \frac{\sqrt{T^3 \left(\frac{1}{M_A} + \frac{1}{M_B}\right)}}{p \,\sigma_{AB}^2 \,\Omega_{D,AB}}$$

M is the molecular weight, p is the pressure, σ_{AB} is the collision diameter, and Ω_{D} , AB is the collision integral and dependent on temperature and intermolecular potential.

Average boundary layer thickness, δ ,

$$\bar{\delta} = \frac{10}{3} \sqrt{\frac{\mu_{mix}L}{\rho U}}$$



PLANETARY MOCVD PROCESS



An Injector MOCVD reaction initiate either surface kinetic or mass transport control. Suppose C_g is the concentration of the bulk gas and C_s is the concentration of reactants at the substrate interface. The concentration of the reactants drops from the bulk to the substrate surface and the corresponding mass flux,

$$I_{gs} = h_g (C_g - C_s)$$

where h_g is the gas mass transfer coefficient, insensitive to variations in temperature.

The flux consumed at the surface $I_s = k_s(C_s)$ where $c_s = -$

where k_{e} is the slowest surface reaction rate constant.

For $k_s >> h_g$, the system dictated by mass controlled , low gas transport rate through the boundary layer limits the rapid surface reaction.

Surface reaction control dominates for $h_g \gg k_s$, the surface reaction is slow even through sufficient reactant gas is available. Additionally, h_g increases with increasing pressure and decreasing temperature and k_s follows the Arrhenius equation.



$$\frac{C_g}{+\frac{k_g}{h_g}}$$



WORKING



- Gas phase kinetics
- Surface phase kinetics
- Each monolayer with atoms positions
- Defects layer by layer quantitatively and qualitatively
- Strain layer by layer
- Surface Roughness
- Lattice Constant etc.





TNL CHEMICAL DATABASE

Precursors for MN Growth

(M = Ga, Al, In) Dopants

- 1. TMM
- 2. DMM
- 1. Cp2 Mg
- 2. Silane

- 3. NH₃
- 4. N₂

Carrier Gases: MN Growth

- 1. N₂
- 2. H₂
- 3. Ar



 $A + B \leftrightarrow AB$

Reaction rates in forward and reverse directions

$$k = AT^n \exp\left(-\right)$$



$\frac{d[AB]}{dt} = \frac{k_1[A][B]^2}{1 + k_2[A] + k_3[B]^{1/2}}$





Total Deposition Rate:

R = A + H + D

A - Adsorption, H – Diffusion, D - Desorption rates

$$\begin{split} A &= Flw & \text{Here, I and w denote length and width of substrate} \\ h_j &= D_0 exp\left(-\frac{E_j}{k_B T}\right) & \text{The characteristic vibration frequency, } D_0 = \\ d_j &= D_0 exp\left(-\frac{E_j^{des}}{k_B T}\right) & \text{with} & E_j^{des} = E_S + nE_n \end{split}$$

 $2k_BT$ h



EXTRACTABLE



Lattice Parameters: 1.

Layer by layer lattice parameter Extraction. Averaging layer by layer lattice constant may produce overall lattice constant of film.

□ The lattice constant can be calibrated with lattice constant with XRD studies.

Lattice constant includes all the strain, defects etc effects.

2. Strain:

□ Averaging layer by layer strain produce overall strain in the film.

The strain can be calibrated with experimental strain.

3. Surface Roughness:

Extract surface roughness as a function of growth time

$$r = \sqrt{\frac{\sum_{i=1}^{N} \sum_{j=1}^{N} [h_{ij} - \overline{h}]^2}{NxN}}$$

Here N is the total number of lattice points, h_{ii} is the height at a given lattice point located at position *i* and *j*, on the lattice and h_{ave} is the average height of all lattice points.

4. Mole fraction:

Extract number of atoms of different constituents layer by layer.

□ Ratio of group-III & V deposited atoms → Molefraction.

5. Defects :

Extract number of interstitials, vacancy etc layer by layer along with dislocation and Stacking Faults





CASE STUDY : **Si/AIN** PLANETARY MOCVD PROCESS





INPUT CONDITIONS

Parameters	Si/AlN	Pre-Nitridated Si/AIN	Parameters	Si/AlN	Pre-
Chamber	300	300			Si/AlN
Chamber Pressure	40	40	Nitridation Temperature (C)	-	1050
(mbar)			Nitridation Time (s)	-	30
Chamber Volume (lits)	1.4	1.4	Surface Energy (eV)	2.0	2.0
Ceiling Height (cm)	1	1	Desorption Barrier (eV)	3.0	3.0
Substrate Temperature (°C)	1050	1050	Schwoebel Barrier (eV)	0.05	0.05
Precursors	TMAI & NH ₃	TMAI & NH ₃	Incorporation Barrier (eV)	0.05	0.05
Precursors Flow Rate (sccm)	105 sccm & 1 slm	105 sccm & 1 slm	Nearest Neighbour (eV)	0.05	0.05
Carrier Gas	H ₂	H ₂	No. of Interactive Elements	1	1
PreNitridation(slm)	-	1	Substrate Dimension (A ²)	50x50	[Unit Cell] ²





Gas Phase Equations

AI(CH3)3 = AICH3 + 2CH3, A=3.5*10^15, AI(CH3)3 + NH3 = AI(CH3)3:NH3, A=3.0*10^12 n=0 AI(CH3)3:NH3 = AI(CH3)3 + NH3, A=5.0*10^10 n=0 AI(CH3)3:NH3 = (CH3)2AI:NH2 + CH4 , A=2.0*10^12 $AI(CH3)3:NH3 + NH3 = (CH3)2AI:NH2 + CH4 + NH3, A=2.0*10^{12}$ 2(CH3)2AI:NH2 = ((CH3)2AI:NH2)2, A=4.0*10^11 n=0 **##Gas to Surface Phase Equations** AI(CH3)3 + space = AI(S) + 3CH3, coll 1.0 AI(CH3)3:NH3 + space = AI(S) + 3CH3 + NH3, coll 1.0 AICH3 + space = AI(S) + CH3, coll 1.0 (CH3)2AI:NH2 + space = AIN(S) + 2CH4, coll 1.0 ((CH3)2AI:NH2)2 + space = 2AIN(S) + 4CH4, coll 1.0

n=0, Ea= 66500
n=0 Ea= 0.0
n=0 Ea= 22000
n=0 Ea= 27000
n=0 Ea= 13000
n=0 Ea= 0.0



OUTPUT



Without Nitridation

Pre-Nitridation for 30 s





DISLOCATIONS PER MONOLAYER





SURFACE ROUGHNESS





OTHER OUTPUT

Parameters	Si/AIN	Pre-Nitridated Si /AIN					
Substrate Thickness (µm)	0.3258	0.3258					
Si3N4 Thickness (nm)	-	3.7817					
Total Deposited Atoms (AIN)	5046178	5233815					
Vacancies (cm ⁻³)	36069	13393					
Total Dislocation Density (cm ⁻³)	2503	1612					
Al Atoms	≈50%	≈50%					
N Atoms	≈50%	≈50%					
Many More							



Patterned Substrate: Selective Epitaxy

TNL-Injector Simulator provides flexibilities to simulate regrowth processes at Atomistic Scale for Selective Epitaxy with capabilities:

- Process Optimization
- > Atomistic growth process for void-semiconductor photonic crystal (PhC)
- > Better understanding of invisible *Physical Phenomenon*
- > Patterned substrates Shapes: Steps, Grooves, Well etc.
- > Epitaxial growth through: MBE, MOVPE/MOCVD
- > Effects of regrowth on **air-hole morphology**
- > Comparison between patterned substrate hole regrown void's dimensions
- > patterned substrate hole: *play a very critical role in the final regrowth*
- > Many More Benefits





MAJOR CHALLENGES: DIMENSIONS OF AIR HOLE





Information Strictly Private and Confidential







Re-growth over Step Patterned Substrate



Regrowth of Si over GaAs Step Pattern Substrate Unit Cells representation

Regrowth of Si over GaAs Step Pattern Substrate Atomistic representation





TNL's tools support advanced and unique licensing models tailored for unique customer needs.

- > ADVANCED LICENSING OPTIONS:
- Term-Based
- Perpetual with Annual Maintenance Cost (AMC)
- TCAD Academic Suite
- 24x7 Technical Support for Academic Institutions



Publications



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